# GNOF: Balanced treatment of electron correlation

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#### Motivation

The need to overcome the drawbacks of currently used DFAs - Efficient description of strongly interacting electrons (molecular dissociation, Mott insulators, etc.)

# Goal

Achieve a more accurate formalism than approximate density functionals, but less computationally demanding than ab initio wavefunction-based methods

# Outline

• Introduction to the Natural Orbital Functional Theory (NOFT)

Reduced Density Matrices (RDMs)

○ N-representability

 $\circ$  2RDM reconstruction in terms of 1RDM: :  $D[n_i, n_j, n_k, n_l]$ 

 $\circ$  Electron-pairing-based NOFs  $\rightarrow$  GNOF

 Born-Oppenheimer Molecular Dynamics (BO-MD) based on an accurate NOF approximation (GNOF)

# **General considerations**

- N-electron system at 0 temperature
- The ground state (GS)
- Non-relativistic spin-free N-electron Hamiltonian:

$$\widehat{H} = \sum_{ik} h_{ik} \widehat{a}_k^{\dagger} \widehat{a}_i + \frac{1}{2} \sum_{ijkl} \langle ij|kl \rangle \widehat{a}_k^{\dagger} \widehat{a}_l^{\dagger} \widehat{a}_j \widehat{a}_i$$

 $\hat{a}_{i}^{\dagger}(\hat{a}_{i})$ : Fermion creation (annihilation) operators associated with the complete orthonormal spin-orbital set { $|i\rangle$ }

• 
$$\left[\widehat{H}, \widehat{S}^2\right] = 0, \left[\widehat{H}, \widehat{S}_z\right] = 0 \implies S \neq 0$$
: GS is a multiplet

• The N-electron system in a mixed state (ensemble) is described by the N-particle density matrix statistical operator:

$${}^{N}\widehat{D} = \sum_{I} \omega_{I} |\Psi_{I}\rangle \langle \Psi_{I}|, \qquad \sum_{I} \omega_{I} = 1, \omega_{I} \ge 0$$

• The energy is an explicitly functional of the 1- and 2-RDMs:

$$E[N, \Gamma, D] = \sum_{I} h_{ik} \Gamma_{ki} + \sum_{ijkl} \langle ij|kl \rangle D_{kl,ij}$$
$$\Gamma_{ki} = \sum_{I} \omega_{I} \langle \Psi_{I}|\hat{a}_{k}^{\dagger} \hat{a}_{i}|\Psi_{I} \rangle , \quad D_{kl,ij} = \frac{1}{2} \sum_{I} \omega_{I} \langle \Psi_{I}|\hat{a}_{k}^{\dagger} \hat{a}_{l}^{\dagger} \hat{a}_{j} \hat{a}_{i}|\Psi_{I} \rangle$$

RDM's general properties: Hermiticity, Nonnegative diagonal elements, ...

# **Universal 1-RDM Functional:**

$$E [N, \Gamma, D] \rightarrow E [N, \Gamma] = \sum_{ik} \mathcal{H}_{ik} \Gamma_{ki} + V_{ee} [N, \Gamma]$$
$$V_{ee} [N, \Gamma] = \min_{E^2 \ni D \rightarrow \Gamma} \left[ \sum_{ijkl} \langle ij|kl \rangle D_{kl,ij} \right] \quad \text{(Gilbert 75, Levy 79, Valone 80)}$$
$$E_{gs} = \min_{\Gamma \in E^1} \left[ \mathcal{H} [N, \Gamma] + V_{ee} [N, \Gamma] \right]$$

1. Limiting situations  $\Rightarrow$  minimization can be exactly solved. In general, constrained search is not appropriate for computation

2.  $V_{ee}$  [N, $\Gamma$ ] has been unreachable so far  $\Rightarrow$  We have to settle for approximations

Spectral decomposition of the 1RDM:  $\Gamma(\mathbf{x}'_1, \mathbf{x}_1) = \sum_i n_i \phi_i(\mathbf{x}'_1) \phi^*_i(\mathbf{x}_1)$ 

 $\{\phi_i(\mathbf{x})\}$ : natural spin-orbitals  $\{n_i\}$ : occupation numbers

- The 1RDM in the natural orbital (NO) representation:  $\Gamma_{ki} = n_i \delta_{ki}$
- Ensemble N-representable 1RDM:  $0 \le n_i \le 1$ ,  $\sum_i n_i = N$
- The universal functional in NO representation:

$$E_{gs} = \min_{\{n_i, \phi_i\} \in E^1} \left\{ \sum_i n_i \mathcal{H}_{ii} + V_{ee} \left[ N, \{n_i, \phi_i\} \right] \right\}$$
$$V_{ee} \left[ N, \{n_i, \phi_i\} \right] = \min_{E^2 \ni D \to \{n_i, \phi_i\}} \left\{ \sum_{ijkl} \langle ij|kl \rangle D_{kl, ij} \right\}$$

Approximate 2RDM:  $D[n_i, n_j, n_k, n_l]$ 

$$E \approx \sum_{i} n_i \mathcal{H}_{ii} + \sum_{ijkl} D[n_i, n_j, n_k, n_l] \langle ij|kl \rangle, \quad E_{gs} = \min_{\{n_i, \phi_i\} \in E^1} E[N, \{n_i, \phi_i\}]$$

- V<sub>ee</sub> will not, in general, be entirely rebuilt
  - An approximate NOF still depends on D

#### The N-representability is twofold:

1) N-representable 1RDM 2) N-representable Functional

 $\mathfrak{D} \to \Gamma : \mathfrak{0} \leq n_i \leq \mathfrak{1}, \ \sum_i n_i = \mathbb{N}$   $\mathfrak{D} \to \mathbb{D} : E[\mathfrak{D}] \to E[\mathbb{N}, \{n_i, \phi_i\}]$ 

• For the 2RDM, a complete set of N-representability conditions that do not depend on higher-order RDMs remains unknown.

Alternative: Use the (2,2)-positivity conditions  $D \ge 0, Q \ge 0, G \ge 0$ 

Singlet State in a two-electron system

• Exact Wavefunction & Energy (Lowdin & Shull 1955)

$$\Psi(\mathbf{x}_{1},\mathbf{x}_{2}) = \frac{1}{\sqrt{2}} \left( \alpha_{1}\beta_{2} - \alpha_{2}\beta_{1} \right) \sum_{p} f_{p} \sqrt{n_{p}} \varphi_{p} \left( \mathbf{r}_{1} \right) \varphi_{p} \left( \mathbf{r}_{2} \right), \ f_{p} = \pm 1$$
$$E\left[ N, \{f_{p}, n_{p}, \varphi_{p}\} \right] = 2 \sum_{p} n_{p} H_{pp} + \sum_{p,q} f_{p} f_{q} \sqrt{n_{q} n_{p}} L_{pq}, \ L_{pq} = \langle pp | qq \rangle$$

Phase Dilemma: # of possible  $\{f_p\}$  combinations is prohibitively large

• NOF: 
$$E(2e^{-}) = 2\sum_{p} n_{p} H_{pp} + n_{1} L_{11} - 2\sum_{p=2}^{l} \sqrt{n_{1} n_{p}} L_{p1} + \sum_{p,q=2} \sqrt{n_{q} n_{p}} L_{pq}$$

Accurate NOF  $\Rightarrow$  Motivation for using **electron-pairs as basic units**!

Electron-pairing-based NOFs for N electrons:

$$D_{pq,rt}^{lphalpha}$$
 ;  $D_{pq,rt}^{lphaeta}$  ;  $D_{pq,rt}^{etaeta}$ 

I. PNOF5, PNOF6, PNOF7 [Phys. Rev. Lett. 119, 063002, 2017]

Static electron correlation

II. GNOF [Phys. Rev. Lett. 127, 233001, 2021]

Static + Dynamic electron correlation

The adjective 'global' is used instead of 'universal': GNOF  $\neq$  Valone's exact.

#### **GNOF for Singlet States**

$$E = E^{intra} + E^{inter}$$

2 
$$E^{inter} = \sum_{p,q=1}^{N_B} I \left\{ n_q n_p \left( 2J_{pq} - K_{pq} \right) + \left( 1 - \delta_{q\Omega^b} \delta_{p\Omega^b} \right) \right.$$
  
 $\left[ n_q^d n_p^d + \Pi \left( n_q^d, n_p^d \right) - \sqrt{n_q h_q n_p h_p} \right] L_{pq}$ 

prime indicates that only the inter-subspace terms are taking into account (p ∈ Ω<sub>f</sub>, q ∈ Ω<sub>g</sub>, f ≠ g).

#### **Dynamic Hole and Occupation Number**

$$h_g^d = h_g \cdot e^{-\left(rac{h_g}{h_c}
ight)^2}, \quad n_p^d = n_p \cdot e^{-\left(rac{h_g}{h_c}
ight)^2}, \quad g = \overline{1, \mathrm{N}/2}, \quad p \in \Omega_g$$



Dynamic hole for  $h_c = 0.02\sqrt{2}$ .

**GNOF** for Multilet States:  $N = N_{II} + N_{I}$ 

 $\langle \hat{S}_z \rangle = \sum \omega_M \langle SM | \hat{S}_z | SM \rangle = 0 \Rightarrow$  the spin-restricted theory can be adopted M = -STotal Spin:  $\langle \hat{S}^2 \rangle = \frac{N_I}{2} \left( \frac{N_I}{2} + 1 \right) \rightarrow S = \frac{N_I}{2}, \quad 2S + 1 = N_I + 1$  $E = E^{intra} + E^{inter}_{HF} + E^{inter}_{sta} + E^{inter}_{dyn}, \qquad N_{\Omega} = N_{II}/2 + N_{I}$  $E^{intra} = \sum_{g=1}^{N_{II}/2} E_g + \sum_{g=N_{II}/2+1}^{N_{\Omega}} H_{gg}, \quad E^{inter}_{HF} = \sum_{p,q=1}^{N_B} ' n_q n_p \left( 2J_{pq} - K_{pq} \right)$  $\boldsymbol{E}_{sta}^{inter} = -\left(\sum_{p=1}^{N_{\Omega}}\sum_{q=N_{\Omega}+1}^{N_{B}} + \sum_{p=N_{\Omega}+1}^{N_{B}}\sum_{q=1}^{N_{\Omega}} + \sum_{p,q=N_{\Omega}+1}^{N_{B}}\right)'\sqrt{n_{q}h_{q}n_{p}h_{p}}L_{pq}$  $-\frac{1}{2}\left(\sum_{p=1}^{N_{\rm II}/2}\sum_{q=N_{\rm II}/2+1}^{N_{\Omega}}+\sum_{p=N_{\rm II}/2+1}^{N_{\Omega}}\sum_{q=1}^{N_{\rm II}/2}\right)'\sqrt{n_qh_qn_ph_p}L_{pq}-\frac{1}{4}\sum_{p,q=N_{\rm II}/2+1}^{N_{\Omega}}K_{pq}$  $E_{dyn}^{inter} = -\left(\sum_{p=1}^{N_{II}/2}\sum_{q=N_{O}+1}^{N_{B}} + \sum_{p=N_{O}+1}^{N_{B}}\sum_{q=1}^{N_{II}/2}\right)'\left(\sqrt{n_{q}^{d}n_{p}^{d}} - n_{q}^{d}n_{p}^{d}\right)L_{pq}$  $+\sum_{p=q=N_{p}+1}^{N_{B}} \prime \left(\sqrt{n_{q}^{d}n_{p}^{d}}+n_{q}^{d}n_{p}^{d}\right) L_{pq}$ 

#### Total energies (Hartrees)

#### Molecules at the Exp. Geom. (cc-pVTZ)

Atoms (aug-cc-pVTZ)

| GS                    | GNOF   | CCSD(T)   | Molecule  | MP2  | GNOF  | CCSD(T)   |
|-----------------------|--|---|---|--|---|---|
| <b>²</b> <i>S</i>     | -0.49983   | -0.49983  | CH₄   | -40.43238  | -40.45533   | -40.45960   |
| <b>1</b> <i>S</i>     | -2.90084   | -2.90084  | C₂H <sub>6</sub>  | -79.67171  | -79.71166   | -79.71789   |
| <b>2</b> <i>S</i>     | -7.45318   | -7.45338  | H₂CO  | -114.34175   | -114.36809  | -114.36928  |
| <b>1</b> <i>S</i>     | -14.63382  | -14.63565   | нсоон   | -189.51455   | -189.54192  | -189.54659  |
| <b>2</b> <sub>P</sub> | -24.60751  | -24.60912   | C <sub>2</sub> FH <sub>3</sub>  | -177.58430   | -177.62610  | -177.62758  |
| <b>з</b> <sub>Р</sub> | -37.79635  | -37.79712   | C₂H₅N   | -133.70022   | -133.73644  | -133.74992  |
| <b>4</b> <i>S</i>     | -54.52947  | -54.53421   | $C_2H_2O_2$   | -227.51149   | -227.54481  | -227.55734  |
| <b>з</b> <sub>Р</sub> | -75.00049  | -74.99967   | CH₃NH₂  | -95.69653  | -95.73376   | -95.73676   |
| <b>2</b> <sub>P</sub> | -99.65391  | -99.65218   | CH3OCH3   | -154.78170   | -154.83262  | -154.83576  |
| <b>1</b> <sub>S</sub> | -128.8442  | -128.8440   | CH₃CH₂OH  | -154.80194   | -154.84906  | -154.85458  |
|                       |  |   |   |  |   |   |
|                       | 0.0012   | -   | MAE (55)  | 30 mHa   | 8 mHa   | -   |
|                       | GS<br>2 <sub>S</sub><br>1 <sub>S</sub><br>2 <sub>S</sub><br>1 <sub>S</sub><br>2 <sub>P</sub><br>3 <sub>P</sub><br>4 <sub>S</sub><br>3 <sub>P</sub><br>2 <sub>P</sub><br>1 <sub>S</sub> | GSGNOF ${}^2S$ -0.49983 ${}^1S$ -2.90084 ${}^2S$ -7.45318 ${}^1S$ -14.63382 ${}^2P$ -24.60751 ${}^3P$ -37.79635 ${}^4S$ -54.52947 ${}^3P$ -75.00049 ${}^2P$ -99.65391 ${}^1S$ -128.8442 | GS       GNOF       CCSD(T) <sup>2</sup> S       -0.49983       -0.49983 <sup>1</sup> S       -2.90084       -2.90084 <sup>2</sup> S       -7.45318       -7.45338 <sup>1</sup> S       -14.63382       -14.63565 <sup>2</sup> P       -24.60751       -24.60912 <sup>3</sup> P       -37.79635       -37.79712 <sup>4</sup> S       -54.52947       -54.53421 <sup>3</sup> P       -75.00049       -74.99967 <sup>2</sup> P       -99.65391       -99.65218 <sup>1</sup> S       -128.8442       -128.8440 | GSGNOFCCSD(T)Molecule $^2S$ -0.49983-0.49983CH4 $^1S$ -2.90084-2.90084C2H6 $^2S$ -7.45318-7.45338H2CO $^1S$ -14.63382-14.63565HCOOH $^2P$ -24.60751-24.60912C2FH3 $^3P$ -37.79635-37.79712C2H5N $^4S$ -54.52947-54.53421C2H2O2 $^3P$ -75.00049-74.99967CH3NH2 $^2P$ -99.65391-99.65218CH3OCH3 $^1S$ -128.8442-128.8440CH3CH2OH | GS       GNOF       CCSD(T)       Molecule       MP2 <sup>2</sup> S       -0.49983       -0.49983       CH <sub>4</sub> -40.43238 <sup>1</sup> S       -2.90084       -2.90084       C <sub>2</sub> H <sub>6</sub> -79.67171 <sup>2</sup> S       -7.45318       -7.45338       H <sub>2</sub> CO       -114.34175 <sup>1</sup> S       -14.63382       -14.63565       HCOOH       -189.51455 <sup>2</sup> P       -24.60751       -24.60912       C <sub>2</sub> FH <sub>3</sub> -177.58430 <sup>3</sup> P       -37.79635       -37.79712       C <sub>2</sub> H <sub>5</sub> N       -133.70022 <sup>4</sup> S       -54.52947       -54.53421       C <sub>2</sub> H <sub>2</sub> O <sub>2</sub> -227.51149 <sup>3</sup> P       -75.00049       -74.99967       CH <sub>3</sub> NH <sub>2</sub> -95.69653 <sup>2</sup> P       -99.65391       -99.65218       CH <sub>3</sub> OCH <sub>3</sub> -154.78170 <sup>1</sup> S       -128.8442       -128.8440       CH <sub>3</sub> CH <sub>2</sub> OH       -154.80194 <sup>1</sup> S       -128.8442       -128.8440       CH <sub>3</sub> CH <sub>2</sub> OH       -154.80194               0.0012       -       MAE (55)       30 mHa | GS       GNOF       CCSD(T)       Molecule       MP2       GNOF <sup>2</sup> S       -0.49983       -0.49983       CH <sub>4</sub> -40.43238       -40.45533 <sup>1</sup> S       -2.90084       -2.90084       C <sub>2</sub> H <sub>6</sub> -79.67171       -79.71166 <sup>2</sup> S       -7.45318       -7.45338       H <sub>2</sub> CO       -114.34175       -114.36809 <sup>1</sup> S       -14.63382       -14.63565       HCOOH       -189.51455       -189.54192 <sup>2</sup> P       -24.60751       -24.60912       C <sub>2</sub> FH <sub>3</sub> -177.58430       -177.62610 <sup>3</sup> P       -37.79635       -37.79712       C <sub>2</sub> H <sub>5</sub> N       -133.70022       -133.73644 <sup>4</sup> S       -54.52947       -54.53421       C <sub>2</sub> H <sub>2</sub> O <sub>2</sub> -227.51149       -227.54481 <sup>3</sup> P       -75.00049       -74.99967       CH <sub>3</sub> NH <sub>2</sub> -95.69653       -95.73376 <sup>2</sup> P       -99.65391       -99.65218       CH <sub>3</sub> OCH <sub>3</sub> -154.78170       -154.83262 <sup>1</sup> S       -128.8442       -128.8440       CH <sub>3</sub> CH <sub>2</sub> OH       -154.80194       -154.84906 <sup>1</sup> S       -128.8442       -128.8440       CH <sub>3</sub> CH <sub>2</sub> OH       -154.80194       -154.84906 <sup>1</sup> S       -10012       - |

# Potential Energy Curves



# Comparison with the experimental data

| Mol   | Mul | $R_{e}$ | $\mathbf{R}_{e}^{exp}$ | $D_e$ | $D_{e}^{exp}$ | $\omega_e$ | $\omega_e^{exp}$ |
|-------|-----|---------|------------------------|-------|---------------|------------|------------------|
| $F_2$ | 1   | 1.35    | 1.41                   | 40.9  | 39.2          | 1212       | 917              |
| $H_2$ | 1   | 0.74    | 0.74                   | 108.6 | 109.5         | 4404       | 4401             |
| BN    | 3   | 1.29    | 1.32                   | 102.3 | 94-133        | 1851       | 1515             |
| CN    | 2   | 1.14    | 1.17                   | 171.6 | 177.4         | 2344       | 2069             |
| CF    | 2   | 1.26    | 1.27                   | 129.0 | 128.7         | 1238       | 1308             |
| CO    | 1   | 1.11    | 1.13                   | 259.6 | 259.3         | 2391       | 2170             |

 $R_e$  (Å),  $D_e$  (kcal/mol), and  $\omega_e$  (cm<sup>-1</sup>)

# PECs for the singlet ground state of the CO



Benchmarking GNOF against FCI in one, two and three dimensions



I. Mitxelena & M. Piris, J. Chem. Phys. 156, 214102 (2022)

Dissociation curves corresponding to a 1D lineal chain of 10 H



# Dissociation curves corresponding to a 2D ring of 10 H



Dissociation curves corresponding to a 2D sheet of 10 H



Dissociation curves corresponding to a 3D pyramid of 10 H



# Dissociation curves corresponding to a 1D lineal chain of 50 H





Alejandro Rivero

# Born-Oppenheimer Molecular Dynamics based on GNOF

 In BOMD calculations, the set of nuclei are propagated according to classical equations of motion, on the potential energy surface (PES) obtained by *on-the-fly* solution of the quantum-mechanical electronic structure problem

$$M_A \ddot{R}_A(t) = -\nabla_A E$$
$$E = E_{nuc} + E_{el} = \sum_{A < B} \frac{Z_A Z_B}{R_{AB}} + E_{el}$$
$$E_{el} = \sum_i n_i \mathcal{H}_{ii} + V_{ee} \left[N, \{n_i\}, \{\phi_i(\mathbf{x})\}\right]$$

- BOMD is, by definition, the true classical dynamics on the BO PES, assuming that the electronic problem is solved exactly (within a given model chemistry) at each time step
- Energy conservation is clearly a desirable feature

MO-LCAO: 
$$\phi_i(\mathbf{x}) = \sum_{\upsilon} C_{\upsilon i} \zeta_{\upsilon}(\mathbf{x}), \ E_{el} = \sum_{\mu \upsilon} \Gamma_{\mu \upsilon} \mathcal{H}_{\mu \upsilon} + \sum_{\mu \upsilon \eta \delta} D_{\mu \eta \upsilon \delta} \langle \mu \eta | \upsilon \delta \rangle$$

The analytic energy gradients with respect to nuclear motion

$$\frac{dE}{dx_A} = \frac{\partial E_{el}}{\partial x_A} + \frac{\partial E_{nuc}}{\partial x_A} + \sum_{\mu j} \frac{\partial E_{el}}{\partial C_{\mu j}} \frac{\partial C_{\mu j}}{\partial x_A} + \sum_i \frac{\partial E_{el}}{\partial n_i} \frac{\partial n_i}{\partial x_A}$$
$$\sum_{\mu j} \frac{\partial E_{el}}{\partial C_{\mu j}} \frac{\partial C_{\mu j}}{\partial x_A} = -\sum_{\mu v} \lambda_{\mu v} \frac{\partial S_{\mu v}}{\partial x_A}, \quad \lambda_{\mu v} = \sum_{ij} C_{\mu j} \lambda_{ji} C_{vi} \quad (\text{density force})$$
$$\sum_i \frac{\partial n_i}{\partial x_A} = 0 \rightarrow \sum_i \frac{\partial E_{el}}{\partial n_i} \frac{\partial n_i}{\partial x_A} = \mu \sum_i \frac{\partial n_i}{\partial x_A} = 0.$$
$$\frac{dE}{dx_A} = \sum_{\mu v} \Gamma_{\mu v} \frac{\partial \mathcal{H}_{\mu v}}{\partial x_A} + \sum_{\mu v \eta \delta} D_{\mu \eta v \delta} \frac{\partial \langle \mu \eta | v \delta \rangle}{\partial x_A} + \frac{\partial E_{nuc}}{\partial x_A} - \sum_{\mu v} \lambda_{\mu v} \frac{\partial S_{\mu v}}{\partial x_A}$$





 $NH(^{3}\Sigma) + NH(^{3}\Sigma) \longrightarrow N_{2}(^{1}\Sigma) + H(^{2}S) + H(^{2}S)$ 







Speed 2x

# $NH(^{3}\Sigma) + NH(^{3}\Sigma) \longrightarrow N_{2}(^{1}\Sigma) + H(^{2}S) + H(^{2}S)$

dt = 1fs $R_{ini} = 10 \text{ Å}$ 





Speed 2x

Π9 Π

### Donostia Natural Orbital Functional Program



# https://github.com/DoNOF/DoNOFsw

- Unconstrained Occupancy Optimization
- Self-consistent iterative diagonalization procedure for NOs
- Geometry optimization (analytic gradients)
- Harmonic vibrational frequencies (numerical Hessian)
- Natural and canonical representations of molecular orbitals
- Ionization potential using extended Koopmans' Theorem (EKT)

More details in <a href="https://donof.readthedocs.io/">https://donof.readthedocs.io/</a>

# **TAKE-HOME MESSAGES**

• Approximate NOF still depends on the 2RDM: Functional N-representability

- GNOF: recover dynamic and non-dynamic electronic correlations
- BO-MD based on GNOF

Thank you for your attention !!!